

VIRIDIFLORIN, AN ISOFLAVONE FROM *TEPHROSIA VIRIDIFLORA**

FEDERICO GÓMEZ, JOSÉ S. CALDERÓN, LEOVIGILDO QUIJANO, MARTHA DOMÍNGUEZ and TIRSO RÍOS

Instituto de Química, Universidad Nacional Autónoma de México (UNAM), Circuito Exterior, Ciudad Universitaria, Coyoacán 04510, México

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Abstract—A new isoflavone, viridiflorin, has been isolated from *Tephrosia viridiflora*. Its structure was established as 4',5,7-trihydroxy-2',5'-dimethoxy-6-prenylisoflavone based on spectral evidence and chemical transformation.

INTRODUCTION

Several reports have indicated that the extract of some species of the genus *Tephrosia* have piscicidal, insecticidal, repellent [1] and anti-cancer properties [2]. As part of our chemical systematic study of the genus *Tephrosia* we have previously investigated *T. madrensis*, *T. watsoniana* and *T. nitens* and isolated a number of novel flavonoids [3, 4]. In cognizance of these results we have undertaken the study of a new species of this genus, *T. viridiflora*, a herbaceous plant endemic to western Mexico.

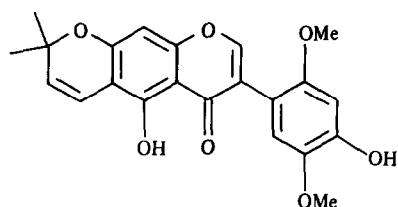
From the roots and aerial parts of *Tephrosia viridiflora* we have isolated a new isoflavone, viridiflorin (**2a**), in

addition to the known elongatin (**1**) [5], the rotenoids villosinol [6] and 11-hydroxytephrosin [7], and the sterols sitosterol and stigmasterol.

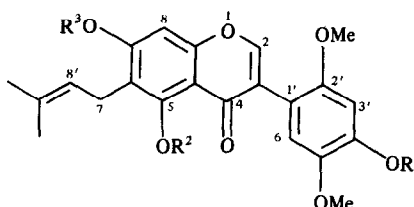
RESULTS AND DISCUSSION

Extraction of the roots and aerial parts of the plant with petrol, ethyl acetate and methanol, followed in each case by CC and prep TLC over silica gel (see Experimental), gave two isoflavonoids. One of them was identified as elongatin (**1**), an isoflavone previously isolated from *T. elongata* E. Mey [5]. The second flavonoid was a new compound which we named viridiflorin.

Viridiflorin (**2a**), $C_{22}H_{22}O_7$ ($[M]^+$ m/z 398) was an isoflavone which showed a strong hydroxyl absorption at 3420 cm^{-1} in the IR spectrum (positive ferric chloride).



1

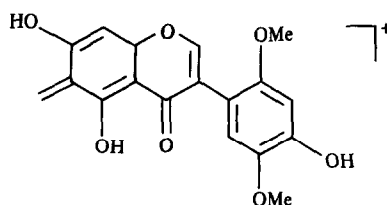


2a $R^1 = R^2 = R^3 = H$

2b $R^1 = R^2 = R^3 = Ac$

2c $R^1 = R^3 = Ac, R^2 = H$

2d $R^1 = R^3 = Me, R^2 = H$



3

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The presence of a hydrogen bonded hydroxyl group at C-5 was indicated by the carbonyl absorption at 1650 cm^{-1} [8] and the typical hydroxyl proton signal at $\delta 13.00$ in the ^1H NMR spectrum (Table 1). The compound also showed a downfield singlet at $\delta 7.8$ characteristic of the C-2 proton of the isoflavone nucleus [9]. Since the ^1H NMR spectrum of viridiflorin (**2a**) exhibited three further singlets at $\delta 6.30$, 6.42 and 6.80 , accounting for three isolated skeletal protons, viridiflorin must be a hexasubstituted isoflavone. The nature of the substituents was evident from the ^1H NMR spectrum, which showed two methoxyl singlets at $\delta 3.70$ and 3.82 and the typical resonances of a prenyl group, observed as two broad singlets at $\delta 1.66$ (3H) and 1.78 (3H), a broad doublet at $\delta 3.30$ (2H) and a broad triplet at $\delta 5.20$ (1H). Two broad singlets interchangeable with deuterated water at $\delta 8.35$ and 10.06 indicated the presence of two extra phenolic hydroxyl groups. As in elongatin (**1**), three substituents, one hydroxyl and two methoxyl groups, can be placed in the B-ring at C-4', C-2' and C-5', respectively, and the remaining prenyl and hydroxyl groups at C-6 and C-7 in the A-ring. Based on the above data, viridiflorin can be represented by structure **2a**.

As expected, the mass spectrum of viridiflorin (**2a**) showed a weak peak at m/z 177 which can be assigned to the fragments $[\text{B}_1 - \text{H}]^+$ and for $[\text{A}_1 - 43]^+$ derived from the RDA process [10]. Besides the $[\text{M}]^+$ at m/z 398 (66%), other intense peaks were observed at m/z 355 $[\text{M} - 43]^+$ (100%) and 343 $[\text{M} - 55]^+$ (88%) due to the fragmentation of the prenyl group. The loss of 55 mass units can give rise to the ion, **3**, which would be in accord with the prenyl group being attached at C-6. The presence of three phenolic groups was confirmed by acetylation of **2a** which provided the corresponding triacetate, **2b**, and the diacetate derivative, **2c**. The chelated hydroxyl group at C-5 was also evident when **2a** was treated with dimethyl sulphate giving the derivative, **2d**. This result is in

accordance with the presence of the bulky prenyl group at C-6 rather than at C-8.

Confirmation of structure **2a** was achieved by cyclo-dehydrogenation with 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ). All the spectral data (IR, UV, NMR and mass spectra) of the obtained product were identical to those of elongatin (**1**).

The isolation of viridiflorin (**2a**) is of particular interest since it is probably the biogenetic precursor of elongatin (**1a**).

EXPERIMENTAL

Tephrosia viridiflora Tellez, was collected in Jalisco, Mexico, ca 60 km, north of Melaque in 1983. A voucher is deposited at the Herbarium of Instituto de Biología (UNAM), Mexico.

Extraction. Air-dried leaves and flowers (270 g) were extracted successively with heptane, EtOAc and MeOH. After evaporation of solvents the green syrups A (17.3 g), B (15.1 g) and C (27.1 g), respectively, were obtained. In the same way, from the air-dried roots (1320 kg), yellow pastes D (18.8 g), E (12.9 g) and F (41.8 g) were obtained.

The heptane extract A (17.3 g) was percolated on a column packed with 170 g Tonsil and eluted with heptane and mixtures of heptane–EtOAc. From the fractions eluted with heptane, a mixture of sitosterol and stigmasterol was obtained.

The EtOAc extract B (15.1 g) was fractionated on Tonsil (150 g) using heptane and EtOAc. The heptane fraction (7.5 g) was chromatographed on silica gel (80 g) using heptane and heptane–Me₂CO mixtures. Fractions eluted with heptane–Me₂CO (19.1) were combined and crystallized from CH₂Cl₂–MeOH giving **1** (198 mg), mp $180\text{--}182^\circ$ (lit. [5] $181\text{--}182^\circ$), identified by IR, ^1H NMR, MS and comparison with an authentic sample. TLC of fractions eluted with heptane–Me₂CO (8.2) afforded 25 mg **2a**.

In the same way, extract D (18.8 g) afforded 136 g of **1**. Extract E (12.9 g) afforded 27 mg **2a** and 36 mg **1**. Finally, from extract F (41.8 g), 10 g **1**, 58.3 mg **2a** and 22 mg of a mixture of villosinol [6] and 11-hydroxytephrosin [7] were obtained.

Viridiflorin (2a) C₂₂H₂₂O₇, colourless needles, mp $220\text{--}222^\circ$ (CH₂Cl₂–hexane). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 204 (12595), 263 (7414), 295 (4815). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3420, 1650, 1515, 1300. EIMS (probe) 70 eV, m/z (rel. int.) 398 $[\text{M}]^+$ (62), 355 $[\text{M} - 43]^+$ (100), 343 $[\text{M} - 55]^+$ (82), 177 $[\text{B}_1 - \text{H}]^+$ and/or $[\text{A}_1 - 43]^+$ (11).

Acetylation of viridiflorin (2a). To a soln of **2a** (16 mg) in pyridine (1 ml) was added Ac₂O (0.5 ml) and the soln heated at 80° for 1.5 hr. The reaction mixture was poured over crushed ice and extracted with CH₂Cl₂, washed with H₂O, dried and evaporated to afford after prep. TLC [hexane–Me₂CO (4.1), $\times 6$] the triacetate, **2b** (10 mg), and the diacetate, **2c** (5 mg).

Viridiflorin triacetate (2b) C₂₈H₂₈O₁₀, colourless needles, mp $200\text{--}202^\circ$. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 215 (8559), 250 (5362), 296 (3825). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 1770, 1650. EIMS (probe) 70 eV, m/z (rel. int.) 524 $[\text{M}]^+$ (17), 482 $[\text{M} - \text{C}_2\text{H}_2\text{O}]^+$ (18), 440 $[\text{M} - \text{C}_4\text{H}_4\text{O}_2]^+$ (25), 397 $[\text{M} - \text{C}_4\text{H}_4\text{O}_2 - \text{C}_2\text{H}_3\text{O}]^+$ (43).

Viridiflorin diacetate (2c) C₂₆H₂₆O₉, colourless needles, mp $140\text{--}144^\circ$. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ) 210 (31206), 262 (18420), 297 (12111). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 1770, 1765, 1645, 1600, 1510. EIMS (probe) 70 eV, m/z (rel. int.) 482 $[\text{M}]^+$ (28), 440 $[\text{M} - \text{C}_2\text{H}_2\text{O}]^+$ (34), 398 $[\text{M} - \text{C}_4\text{H}_4\text{O}_2]^+$ (12), 397 $[\text{M} - \text{C}_4\text{H}_5\text{O}_2]^+$ (52), 385 $[\text{M} - \text{C}_6\text{H}_9\text{O}]^+$ (20), 343 $[\text{M} - \text{C}_8\text{H}_{11}\text{O}_2]^+$ (41), 43 (100).

Methylation of viridiflorin (2a). A mixture of **2a** (24 mg) dry K₂CO₃ (30 mg) and Me₂SO₄ (1 ml) in dry Me₂CO (20 ml) was refluxed for 3 hr and worked-up as usual. Purification by TLC [hexane–Me₂CO (4.1) $\times 6$] of the reaction residue afforded **2d** (9 mg), mp $193\text{--}195^\circ$. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (ϵ) 213 (16111), 263 (11389), 290 (7543). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 1650, 1590, 1510, 1300. EIMS

Table 1 ^1H NMR data of viridiflorin (**2a**) and its derivatives (80 MHz, deuteriochloroform, TMS as int. standard)

	2a *	2b	2c	2d
H-2	7.8 s	7.8 s	7.9 s	7.8 s
H-8	6.42 s	6.7 s	6.72 s	6.38 s
H-3'	6.6 s	6.93 s	6.72 s	6.6 s
H-6'	6.8 s	7.18 s	6.97 s	6.87 s
H-7†	3.3 d	3.28 d	3.31 d	3.35 d
H-8‡	5.2 t	5.03 t	5.13 t	5.09 t
gem-Me ₂	1.65 s	1.69 s	1.69 s	1.67 s
	1.77 s	1.79 s	1.77 s	1.78 s
OH	13.0 s	—	13.02 s	12.9 s
	10.07 s			
	8.35 s			
OMe	3.7 s	3.7 s	3.73 s	3.76 s
	3.81 s	3.8 s	3.8 s	3.84 s
				3.89 s
				3.92 s
OAc	—	2.33 s	2.35 s	—
		2.35 s		
		2.42 s		

* Run in deuteriochloroform–DMSO

† d, $J_{7,8} = 7\text{ Hz}$

‡ t, $J_{8,7} = 7\text{ Hz}$

(probe) 70 eV, m/z (rel int) 426 $[M]^+$ (51), 395 $[M - 31]^+$ (22), 383 $[M - 43]^+$ (98), 371 $[M - 55]^+$ (100), 191 $[C_{10}H_7O_4]^+$ (18)

Conversion of viridiflorin (2a) to elongatin (1) A soln of 34 mg 2a and 30 mg DDQ in 40 ml dry C_6H_6 was refluxed for 2 hr After usual work-up followed by prep TLC [hexane- Me_2CO (3:2) \times 3] 1 was obtained identical in all respects (TLC, IR, 1H NMR, mp) with authentic 1

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